

Copper-Catalyzed Intramolecular Cyclization of *N*-Propargyl-Adenine: Synthesis of Purine-Fused Tricyclics

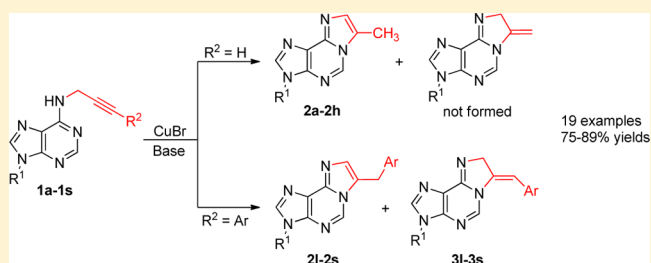
Ren-Long Li,[†] Lei Liang,[†] Ming-Sheng Xie,[†] Gui-Rong Qu,^{*,†} Hong-Ying Niu,[‡] and Hai-Ming Guo^{*,†}

[†]Collaborative Innovation Center of Henan Province for Green Manufacturing of Fine Chemicals, Key Laboratory of Green Chemical Media and Reactions, Ministry of Education, School of Chemistry and Chemical Engineering, Henan Normal University, Xinxiang, Henan 453007, PR China

[‡]School of Chemistry and Chemical Engineering, Henan Institute of Science and Technology, Xinxiang, Henan 453003, PR China

Supporting Information

ABSTRACT: A novel protocol to construct fluorescent purine-fused tricyclic products via intramolecular cyclization of *N*-propargyl-adenine has been developed. With CuBr as the catalyst, a series of purine-fused tricyclic products were obtained in good to excellent yields (19 examples, 75–89% yields). When R₂ was a hydrogen atom in *N*-propargyl-adenines, the reactions only afforded the endocyclic double bond products. When R₂ was an aryl group, the electron-donating groups favored the endocyclic double bond products, while the electron-withdrawing groups favored the exocyclic double bond products.



Purine-fused tricyclic and polycyclic derivatives are an important class of nitrogen-containing heterocycles that have attracted considerable attention in biochemistry, molecular biology, and toxicology.¹ For example, purine-fused tricyclic 1,*N*⁶-ethenoadenine (Figure 1) has been widely studied

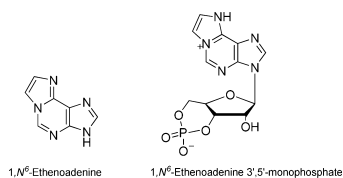
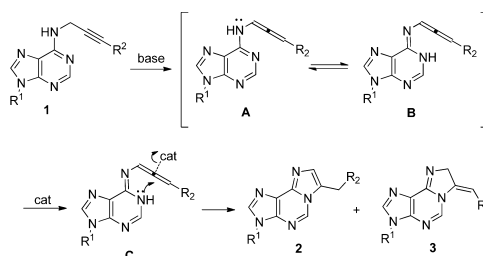


Figure 1. Selected examples of purine-fused tricyclics exhibited high fluorescence properties.

because of their excellent fluorescence properties.² And 1,*N*⁶-ethenoadenine 3',5'-monophosphate is a highly fluorescent probe.^{2f,g} Thus, chemists have tried to synthesize different purine-fused polycyclics and studied their fluorescence properties and biological activities.³ The classical approach for the synthesis of purine-fused polycyclics is via the reaction of haloacetaldehyde with adenine or adenosine.⁴ Although great endeavor has been devoted to the synthesis of purine-fused polycyclics, the structure diversity of purine-fused polycyclic derivatives is still very limited. Therefore, searching for a new approach for the synthesis of purine-fused polycyclic with structure diversity is highly desirable.

We proposed the purine-fused tricyclics could be constructed as follows (Scheme 1): allenamine intermediate **A** might be generated from substrate **1** and a base,^{5,6} and then intermediate **B**, the tautomer of **A**, could be activated by a catalyst, and the

Scheme 1. Synthesis of Purine-Fused Tricyclic Derivatives Starting from *N*-Propargyl-adenine

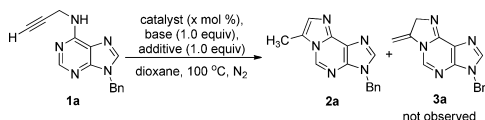


nucleophilic attack of N1 to allenamine through **C** would lead to the formation of two purine-fused tricyclic products **2** and **3**. In the context of ongoing projects on the modification of purine analogues,^{7,8} herein we report a useful and facile alternative process for the synthesis of fluorescent purine-fused tricyclic derivatives by using substituted propargyl-adenine compounds as starting material.⁹

Initially, we conducted our study by treating *N*-propargyl-9-benzyl-adenine **1a** with CuBr in the presence of CsCO₃ in dioxane (Table 1). To our delight, the cyclization reaction occurred, and purine-fused tricyclic product **2a** was formed, though the yield was not satisfied (entry 1). Next, some quaternary ammonium salts such as *n*-Bu₄NBr, Me₄NBr, or Me₄NCl were added into the reaction (entries 2–5).¹⁰ The results showed that *n*-Bu₄NBr was the best one affording product **2a** with 86% yield (entry 4). When the catalyst loading

Received: January 24, 2014

Published: March 28, 2014

Table 1. Optimization of the Reaction Conditions^a


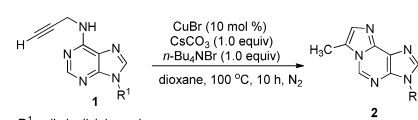
entry	catalyst (x)	base	additive	yield (%) ^b
1	CuBr (20)	Cs ₂ CO ₃	—	55
2	CuBr (20)	Cs ₂ CO ₃	<i>n</i> -Bu ₄ NI	48
3	CuBr (20)	Cs ₂ CO ₃	Me ₄ NBr	25
4	CuBr (20)	Cs ₂ CO ₃	Me ₄ NCl	32
5	CuBr (20)	Cs ₂ CO ₃	<i>n</i> -Bu ₄ NBr	86
6	CuBr (10)	Cs ₂ CO ₃	<i>n</i> -Bu ₄ NBr	87
7 ^c	CuBr (10)	Cs ₂ CO ₃	<i>n</i> -Bu ₄ NBr	52
8 ^d	CuBr (10)	Cs ₂ CO ₃	<i>n</i> -Bu ₄ NBr	60
9 ^e	CuBr (10)	Cs ₂ CO ₃	<i>n</i> -Bu ₄ NBr	72
10	CuBr (10)	Na ₂ CO ₃	<i>n</i> -Bu ₄ NBr	60
11	CuBr (10)	K ₂ CO ₃	<i>n</i> -Bu ₄ NBr	72
12	CuBr (10)	<i>t</i> -BuOK	<i>n</i> -Bu ₄ NBr	trace
13	CuCl (10)	Cs ₂ CO ₃	<i>n</i> -Bu ₄ NBr	68
14	CuI (10)	Cs ₂ CO ₃	<i>n</i> -Bu ₄ NBr	79
15	CuBr ₂ (10)	Cs ₂ CO ₃	<i>n</i> -Bu ₄ NBr	trace

^aThe reactions were carried out with **1a** (0.3 mmol), catalyst, base (0.3 mmol), solvent (2.0 mL), additive (1.0 equiv) in a Schlenk tube at 100 °C for 10 h under N₂ atmosphere. ^bIsolated yield. ^cToluene was used as the solvent. ^dMeCN was used as the solvent. ^eDMF was used as the solvent.

was lowered to 10 mol %, the reaction also proceeded well, giving the product **2a** with 87% yield (entry 6). The screening of solvents showed that dioxane was the suitable one (entries 6–9). Next, a range of bases were examined, and Cs₂CO₃ emerged as the best choice (entries 10–12). Subsequently, several copper salts were probed (entries 13–15), and for CuCl₂, Cu(OAc)₂, Cu(OTf)₂, CuSO₄, trace amounts of products were obtained. It was found that Cu(I) exhibited better catalytic activity than Cu(II), and CuBr was the most suitable copper source. Finally, the optimal conditions were determined to be 10 mol % of CuBr, 1.0 equiv of CsCO₃, and 1 equiv of *n*-Bu₄NBr in dioxane (entry 6).

With the optimized conditions (Table 1, entry 6) in hand, the scope of this cyclization reaction was investigated. As shown in Table 2, a series of N9-substituted substrates, including alkyl, benzyl, allyl, and sugar groups, were subjected to the optimized reaction conditions. The results showed that all the substrates could be transformed into the purine-fused tricyclics in good yields (**2a–2h**, 76–88% yields). Notably, the allyl substituted **1e** and glycosidic substituted **1h** are suitable substrates for the reactions (78% yield for **2e**, 87% yield for **2h**), which offers an ideal opportunity for further synthetic manipulation.

Subsequently, various *N*-propargyl-adenines **1i–1s** with different aryl groups at the terminal position of alkenyl part were synthesized via Sonogashira coupling reactions. As shown in Table 3, when Ar was phenyl group, the corresponding purine-fused tricyclic products **2i** and **3i** were obtained in 88% total yield, in which the endocyclic double bond product **2i** was the major form (entry 1). Next, a series of electron-donating groups at the benzene ring in the alkenyl part of 9-benzyl-adenines were evaluated (entries 2–7). When *ortho*- or *meta*-methyl substituted substrates were explored, the reactions proceeded well, affording purine-fused tricyclic products in excellent yields, in which the endocyclic double bond products **2j** and **2k** were the major form (entries 2–3). Meanwhile, the

Table 2. Effect of N9-Substituent in the *N*-Propargyl-adenine^a


entry	Substrate	Product	yield (%) ^b
1			87
2			88
3			85
4			84
5			78
6			80
7			76
8			87

^aThe reactions were carried out with **1** (0.3 mmol), CuBr (10 mol %), CsCO₃ (0.3 mmol), dioxane (2.0 mL), *n*-Bu₄NBr (0.3 mmol) in a Schlenk tube at 100 °C for 10 h under N₂ atmosphere. ^bIsolated yield.

3,5-dimethylphenyl substituted substrate **1l** could also be tolerated in the reaction, giving the endocyclic double bond product **2l** as the major form (entry 4). Furthermore, several alkoxy groups (OCH₃, OEt) substituted substrates were tested, and the cyclization reactions worked well, affording the corresponding purine-fused tricyclic products in 85–88% yields, in which the endocyclic double bond products **2m**, **2n**, **2o** were the major form, respectively (entries 5–7). Then, several electron-withdrawing groups at the benzene ring in the alkenyl part of 9-benzyl-adenines were examined (entries 8–10). To our delight, these substrates (**1p–1r**) could afford the cyclization products in good yields (75–88% yields). Contrarily, the exocyclic double bond products **3p**, **3q**, **3r** were the

Table 3. Effect of Aryl Groups at the Terminal Position of Alkynyl Part in *N*-Propargyl-adenine^a

entry	1	Product 2	Product 3	yield (%) ^b	2:3
1	1i			88	99:1
2	1j			89	92:8
3	1k			87	91:9
4	1l			79	87:13
5	1m			85	86:14
6	1n			85	88:12
7	1o			88	92:8
8 ^c	1p			81	47:53
9	1q			88	45:55
10	1r			75	43:57
11	1s			77	45:55

^aThe reactions were carried out with **1** (0.3 mmol), CuBr (10 mol %), CsCO₃ (0.15 mmol), dioxane (2.0 mL), *n*-Bu₄NBr (0.3 mmol) in a Schlenk tube at 100 °C for 10 h under N₂ atmosphere. ^bIsolated yield. ^cThe *Z*-configuration of product **3p** was confirmed by NOESY spectrum.

major form (entries 8–10). Particularly noteworthy is that the terminal alkynes with heterocyclic connections could also

survive this reaction well (entry 11). Take product **3p** as the representative product; the structure of **3p** with exocyclic *Z*-form double bond was confirmed by the ¹H, ¹³C, and NOESY spectra (see Supporting Information for details). It should be noted that the isomerization of products **2** and **3** could not occur under the optimized reaction conditions. Meanwhile, the endocyclic double bond product **2** and the exocyclic double bond product **3** could be separated through column chromatography. In addition, the fluorescence properties of selected products (**2i**, **2j**, **2m**) were studied, and the results indicated that these products showed moderate fluorescence intensity (see Supporting Information for details).

In conclusion, we have developed a new approach to construct purine-fused tricyclic products via copper-catalyzed intramolecular cyclization of *N*-propargyl-adenine. With CuBr as the catalyst, a series of purine-fused tricyclic products were obtained in good to excellent yields (19 examples, 75–89% yields). Meanwhile, the cyclization reactions only afforded the endocyclic double bond products with *N*-propargyl-adenines as the starting materials. As for the electronic effects of the substituents at the benzene ring in the alkynyl part, the electron-donating groups favored the endocyclic double bond products. Contrarily, the electron-withdrawing groups favored the exocyclic double bond products.

EXPERIMENTAL SECTION

General Methods. Melting points were recorded with a melting point microapparatus and are uncorrected. NMR spectra were recorded with a 400 NMR spectrometer for ¹H NMR, 100 MHz for ¹³C NMR. Chemical shifts δ were given in ppm relative to tetramethylsilane in CDCl₃ or (CD₃)₂SO for ¹H and ¹³C NMR. High resolution mass spectra are taken using Q-TOF system, with electrospray ionization (ESI) as the ionization method used for the HRMS measurement. All reactions were set up in air (with no use of a glovebox) and carried out in a sealed pressure tube.

Typical Procedures. *For the Synthesis of 2a–2h.* In a sealed tube, **1a–1h** (0.3 mmol), CuBr (4.3 mg, 0.03 mmol), CsCO₃ (97.7 mg, 0.3 mmol), *n*-Bu₄NBr (96.7 mg, 0.3 mmol), and then dioxane (2.0 mL) were added, and the reaction mixture was heated at 100 °C for 10 h. After cooling down to room temperature, the mixture was diluted with brine (3.0 mL) and extracted with EtOAc (3 × 3.0 mL). Then, the organic layer was collected and dried over anhydrous Na₂SO₄, filtered and concentrated under reduced pressure. The crude material was purified by column chromatography on silica gel (eluting with Pet/EtOAc mixtures) to give the desired products.

For the Synthesis of 2i–2s and 3i–3s. In a sealed tube, **1i–1s** (0.3 mmol), CuBr (4.3 mg, 0.03 mmol), CsCO₃ (48.4 mg, 0.15 mmol), *n*-Bu₄NBr (96.7 mg, 0.3 mmol), and then dioxane (2.0 mL) were added, and the reaction mixture was heated at 100 °C for 10 h. After cooling down to room temperature, the mixture was diluted with brine (3.0 mL) and extracted with EtOAc (3 × 3.0 mL). Then, the organic layer was collected and dried over anhydrous Na₂SO₄, filtered and concentrated under reduced pressure. The crude material was purified by column chromatography on silica gel (eluting with Pet/EtOAc mixtures) to give the desired products.

For the Synthesis of Starting Materials 1a–1h.^{11,12} 6-Chloro-9-substituent-purine (3.0 mmol), propargylic amide (3.6 mmol), Et₃N (3.6 mmol), and ethanol (10.0 mL) were added in a 50 mL tube. The tube was refluxed in a 90 °C bath and stirred for 6 h. After cooling down to room temperature, the mixture was diluted with water and then extracted with EtOAc. The extracts were combined, washed with brine, and then dried over anhydrous Na₂SO₄. The crude material was purified by column chromatography on silica gel (eluting with Pet/EtOAc mixtures) to give the 6-propargylic-9-substituent-adenine.

For the Synthesis of Starting Materials 1i–1s.^{11,12} 6-Propargylic-9-benzyl-adenine (1.5 mmol), iodobenzene (2.25 mmol), Pd(PPh₃)₂Cl₂ (5 mol %), CuI (10 mol %), and Et₃N (5.0 mL) were added in a 50

mL tube. The tube was stirred for 12 h at 35 °C. After cooling down to room temperature, the mixture was diluted with water and then extracted with EtOAc. The extracts were combined, washed with brine, and then dried over anhydrous Na₂SO₄. The crude material was purified by column chromatography on silica gel (eluting with Pet/EtOAc mixtures) to give the 6-proargylic-9-substituent-adenine.

3-Benzyl-7-methyl-3H-imidazo[2,1-*i*]purine (2a). Yellow solid (68.7 mg, 87%): mp 195–197 °C; ¹H NMR (400 MHz, (CD₃)₂SO) δ 8.97 (s, 1H), 8.39 (s, 1H), 7.29–7.22 (m, 6H), 5.50 (s, 2H), 3.34 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 135.5, 133.3, 131.0, 128.8, 128.1, 127.4, 47.6, 9.1; HRMS calcd for C₁₅H₁₄N₅ [M + H⁺] 264.1244, found 264.1236; IR (KBr) ν_{max} 3054, 2975, 2924, 1637, 1371 cm⁻¹.

3-Ethyl-7-methyl-3H-imidazo[2,1-*i*]purine (2b). Brown oil (52.5 mg, 88%): ¹H NMR (400 MHz, CDCl₃) δ 8.58 (s, 1H), 7.96 (s, 1H), 7.37 (s, 1H), 7.36 (q, J = 7.2 Hz, 2H), 2.59 (s, 3H), 1.59 (q, J = 7.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 139.7, 138.6, 133.0, 131.0, 123.7, 118.9, 39.4, 15.7, 9.2; HRMS calcd for C₁₀H₁₁N₅Na [M + Na⁺] 224.0907, found 224.0899; IR (KBr) ν_{max} 2961, 2874, 1679, 1371 cm⁻¹.

3-Isopropyl-7-methyl-3H-imidazo[2,1-*i*]purine (2c). Brown oil (54.9 mg, 85%): ¹H NMR (400 MHz, CDCl₃) δ 8.56 (s, 1H), 7.99 (s, 1H), 7.32 (s, 1H), 4.94–4.87 (m, 1H), 2.56 (s, 3H), 1.64 (d, J = 6.8 Hz, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 138.3, 137.8, 132.8, 130.8, 123.8, 118.9, 47.7, 22.8, 9.2; HRMS calcd for C₁₁H₁₄N₅ [M + H⁺] 216.1244, found 216.1237; IR (KBr) ν_{max} 2969, 2933, 2877, 1638, 1370 cm⁻¹.

3-(Sec-butyl)-7-methyl-3H-imidazo[2,1-*i*]purine (2d). Brown oil (57.7 mg, 84%): ¹H NMR (400 MHz, CDCl₃) δ 8.79 (s, 1H), 7.96 (s, 1H), 7.51 (s, 1H), 4.68–4.60 (m, 1H), 2.51 (s, 3H), 2.07–1.90 (m, 2H), 1.62 (d, J = 6.8 Hz, 3H), 0.85 (t, J = 5.4 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 142.8, 140.9, 139.4, 138.9, 134.3, 123.0, 107.4, 53.6, 29.8, 20.8, 14.1, 10.6; HRMS calcd for C₁₂H₁₆N₅ [M + H⁺] 230.1400, found 230.1401; IR (KBr) ν_{max} 2969, 2933, 2877, 1638, 1370 cm⁻¹.

3-Allyl-7-methyl-3H-imidazo[2,1-*i*]purine (2e). White solid (49.9 mg, 78%): mp 101–103 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.77 (s, 1H), 8.21 (s, 1H), 7.54 (s, 1H), 6.12–6.03 (m, 1H), 5.35 (d, J = 10.4 Hz, 1H), 5.23 (d, J = 16.8 Hz, 1H), 5.02–4.98 (m, 2H), 2.68 (s, 3H); ¹³C NMR (100 MHz, (CD₃)₂SO) δ 141.3, 138.6, 135.4, 133.7, 130.3, 122.7, 120.3, 117.8, 108.6, 46.0, 9.2; HRMS calcd for C₁₁H₁₂N₅ [M + H⁺] 214.1087, found 214.1083; IR (KBr) ν_{max} 3077, 2920, 1639, 1369 cm⁻¹.

7-Methyl-3-(4-methylbenzyl)-3H-imidazo[2,1-*i*]purine (2f). Yellow solid (66.5 mg, 80%): mp 193–195 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.71 (s, 1H), 7.91 (s, 1H), 7.41 (s, 1H), 7.16 (dd, J = 14.4, 8.0 Hz, 4H), 5.40 (s, 2H), 2.49 (s, 3H), 2.32 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 143.6, 141.0, 140.2, 139.0, 138.1, 134.5, 132.4, 129.5, 127.5, 123.0, 107.0, 47.4, 21.0, 14.2; HRMS calcd for C₁₆H₁₆N₅ [M + H⁺] 278.1400, found 278.1402; IR (KBr) ν_{max} 3093, 2944, 2916, 1639, 746 cm⁻¹.

3-(2-Chlorobenzyl)-7-methyl-3H-imidazo[2,1-*i*]purine (2g). Brown solid (67.7 mg, 76%): mp 161–163 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.59 (s, 1H), 8.01 (s, 1H), 7.42 (d, J = 8.4 Hz, 1H), 7.36 (s, 1H), 7.29–7.13 (m, 3H), 5.60 (s, 2H), 2.57 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 140.5, 138.8, 134.7, 133.4, 133.3, 133.2, 131.2, 129.9, 129.8, 129.6, 127.3, 123.5, 119.1, 45.4, 9.3; HRMS calcd for C₁₅H₁₃ClN₅ [M + H⁺] 298.0854, found 298.0850; IR (KBr) ν_{max} 3073, 2991, 2921, 1640, 1370, 748 cm⁻¹.

3-((3a*S*,4*S*,6*S*,6a*S*)-2,2-Dimethyl-6-((trityloxy)methyl)-tetrahydrofuro[3,4-*d*]1,3-dioxol-4-yl)-7-methyl-3H-imidazo[2,1-*i*]purine (2h). Brown oil (153.3 mg, 87%): ¹H NMR (400 MHz, CDCl₃) δ 8.26 (s, 1H), 8.01 (s, 1H), 7.38 (s, 1H), 7.34 (s, 3H), 7.32 (s, 3H), 7.17 (s, 1H), 7.15 (s, 3H), 7.13 (s, 2H), 7.11 (s, 1H), 7.09 (s, 1H), 7.07 (s, 1H), 6.20 (d, J = 2.4 Hz, 3H), 5.38 (dd, J = 6.0, 2.0 Hz, 1H), 4.98 (q, J = 3.2 Hz, 1H), 4.56–4.53 (m, 1H), 3.36–3.27 (m, 1H), 2.53 (s, 3H), 2.03 (s, 1H), 1.63 (s, 3H), 1.38 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 143.3, 137.3, 133.0, 131.3, 128.5, 127.6, 126.9, 114.2, 91.3, 86.7, 86.3, 84.4, 81.9, 77.3, 76.9, 76.6, 63.9, 27.1, 25.4, 9.2; HRMS calcd for C₃₅H₃₄N₅O₄ [M + H⁺] 588.2605, found 588.2603; IR (KBr) ν_{max} 3056, 3032, 2987, 2933, 1641, 1367, 1212, 1077 cm⁻¹.

3,7-Dibenzyl-3H-imidazo[2,1-*i*]purine (2i). Brown oil (88.6 mg, 87.1%): ¹H NMR (400 MHz, CDCl₃) δ 8.45 (s, 1H), 7.93 (s, 1H), 7.46 (s, 1H), 7.31–7.23 (m, 8H), 7.21 (s, 1H), 7.19 (s, 1H), 5.40 (s, 2H), 4.31 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 140.3, 138.8, 136.1, 135.4, 133.6, 132.3, 128.9, 128.3, 128.1, 127.4, 127.1, 47.6, 30.4; HRMS calcd for C₂₁H₁₈N₅ [M + H⁺] 340.1557, found 340.1550; IR (KBr) ν_{max} 3100, 3029, 2923, 1640, 1374 cm⁻¹.

3-Benzyl-7-benzylidene-7,8-dihydro-3H-imidazo[2,1-*i*]purine (3i). Light yellow solid (0.9 mg, 0.9%): mp 141–143 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.64 (s, 1H), 7.91 (s, 1H), 7.34–7.23 (m, 8H), 7.25–7.20 (m, 3H), 5.44 (s, 1H), 4.12 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 148.0, 141.3, 140.3, 139.0, 135.5, 134.8, 129.1, 129.0, 128.5, 128.3, 127.5, 126.4, 123.3, 107.6, 47.7, 35.6; HRMS calcd for C₂₁H₁₈N₅ [M + H⁺] 340.1557, found 340.1561; IR (KBr) ν_{max} 3061, 3026, 2923, 1638, 1372, 694 cm⁻¹.

3-Benzyl-7-(2-methylbenzyl)-3H-imidazo[2,1-*i*]purine (2j). Brown oil (86.7 mg, 81.9%): ¹H NMR (400 MHz, CDCl₃) δ 8.50 (s, 1H), 7.97 (s, 1H), 7.35–7.27 (m, 6H), 7.25–7.19 (m, 2H), 7.13 (d, J = 7.2 Hz, 1H), 6.99 (d, J = 7.6 Hz, 1H), 5.46 (s, 2H), 4.23 (s, 2H), 2.38 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 140.4, 136.4, 135.4, 134.2, 133.7, 132.6, 130.7, 129.0, 128.6, 128.4, 127.6, 127.3, 126.5, 123.7, 121.5, 47.6, 28.2, 19.5; HRMS calcd for C₂₂H₁₉N₅Na [M + Na⁺] 376.1533, found 376.1529; IR (KBr) ν_{max} 3065, 2924, 1641, 1376, 741 cm⁻¹.

3-Benzyl-7-(2-methylbenzylidene)-7,8-dihydro-3H-imidazo[2,1-*i*]purine (3j). Brown oil (7.5 mg, 7.1%): ¹H NMR (400 MHz, CDCl₃) δ 8.54 (s, 1H), 7.86 (s, 1H), 7.26 (d, J = 7.2 Hz, 2H), 7.19 (d, J = 5.2 Hz, 4H), 7.12 (s, 3H), 6.95 (s, 1H), 5.37 (s, 2H), 4.13 (s, 1H) 2.25 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 147.8, 141.3, 140.3, 139.2, 137.2, 136.6, 135.5, 134.8, 130.3, 130.0, 129.0, 128.4, 127.5, 126.7, 126.1, 123.3, 107.6, 47.7, 33.4, 19.5; HRMS calcd for C₂₂H₂₀N₅ [M + H⁺] 354.1713, found 354.1706; IR (KBr) ν_{max} 3084, 3030, 2944, 1640, 1375, 695 cm⁻¹.

3-Benzyl-7-(3-methylbenzyl)-3H-imidazo[2,1-*i*]purine (2k). Brown oil (83.9 mg, 79.2%): ¹H NMR (400 MHz, CDCl₃) δ 8.45 (s, 1H), 7.94 (s, 1H), 7.46 (s, 1H), 7.31–7.16 (m, 6H), 7.05 (d, J = 7.6 Hz, 1H), 7.00 (d, J = 6.0 Hz, 2H), 5.41 (s, 2H), 4.27 (s, 2H), 2.28 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 141.7, 140.4, 138.9, 138.7, 136.0, 135.4, 133.8, 132.1, 129.0, 128.9, 128.8, 128.4, 127.9, 127.5, 125.2, 123.4, 122.1, 47.7, 30.3, 21.3; HRMS calcd for C₂₂H₁₉N₅Na [M + Na⁺] 376.1533, found 376.1528; IR (KBr) ν_{max} 3135, 3029, 2941, 1640, 1371, 766 cm⁻¹.

3-Benzyl-7-(3-methylbenzylidene)-7,8-dihydro-3H-imidazo[2,1-*i*]purine (3k). Orange solid (8.3 mg, 7.8%): mp 138–140 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.65 (s, 1H), 7.93 (s, 1H), 7.34–7.15 (m, 9H), 7.05 (d, J = 7.2 Hz, 1H), 5.45 (s, 2H), 4.12 (s, 2H), 2.34 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 148.1, 141.2, 140.3, 138.9, 138.1, 135.5, 134.9, 129.9, 129.0, 128.3, 127.5, 127.1, 126.1, 123.2, 107.6, 47.7, 35.5, 21.4; HRMS calcd for C₂₂H₂₀N₅ [M + H⁺] 354.1713, found 354.1705; IR (KBr) ν_{max} 3077, 3029, 2908, 1643, 1378, 699 cm⁻¹.

3-Benzyl-7-(3,5-dimethylbenzyl)-3H-imidazo[2,1-*i*]purine (2l). Brown oil (75.7 mg, 68.7%): ¹H NMR (400 MHz, CDCl₃) δ 8.48 (s, 1H), 7.95 (s, 1H), 7.45 (s, 1H), 7.34–7.31 (m, 3H), 7.27–7.25 (m, 2H), 7.25 (s, 1H), 6.88 (s, 1H), 6.82 (s, 2H), 5.43 (s, 2H), 4.25 (s, 2H), 2.25 (s, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 140.3, 138.5, 136.0, 135.4, 133.9, 132.2, 129.0, 128.8, 128.4, 127.5, 126.0, 123.6, 122.1, 47.7, 30.3, 21.2; HRMS calcd for C₂₃H₂₂N₅ [M + H⁺] 368.1870, found 368.1869; IR (KBr) ν_{max} 3125, 3101, 3021, 1641, 1372, 783 cm⁻¹.

3-Benzyl-7-(3,5-dimethylbenzylidene)-7,8-dihydro-3H-imidazo[2,1-*i*]purine (3l). Yellow solid (11.3 mg, 10.3%): mp 152–154 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.64 (s, 1H), 7.92 (s, 1H), 7.34–7.30 (m, 3H), 7.27 (d, J = 2.4 Hz, 1H), 7.25 (s, 1H), 7.22 (s, 1H), 6.98 (s, 1H), 6.87 (s, 1H), 5.44 (s, 2H), 4.11 (s, 2H), 2.29 (s, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 148.2, 141.2, 140.3, 139.1, 138.9, 137.9, 135.5, 134.8, 129.0, 128.4, 128.0, 127.5, 126.9, 123.3, 107.5, 47.7, 35.4, 21.2; HRMS calcd for C₂₃H₂₂N₅ [M + H⁺] 368.1870, found 368.1866; IR (KBr) ν_{max} 3125, 3082, 3015, 1636, 1375, 779, 723 cm⁻¹.

3-Benzyl-7-(2-methoxybenzyl)-3H-imidazo[2,1-*i*]purine (2m). Yellow oil (81.0 mg, 73.1%): ¹H NMR (400 MHz, CDCl₃) δ 8.61 (s,

1H), 7.92 (s, 1H), 7.43 (s, 1H), 7.30 (s, 3H), 7.25–7.19 (m, 3H), 7.00 (d, $J = 7.2$ Hz, 1H), 6.84 (t, $J = 7.6$ Hz, 1H), 5.43 (s, 2H), 4.29 (s, 2H), 3.88 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 156.9, 140.3, 138.9, 135.5, 134.2, 132.0, 129.6, 129.0, 128.4, 127.5, 124.7, 123.5, 122.4, 120.9, 110.6, 55.5, 47.7, 23.8; HRMS calcd for $\text{C}_{22}\text{H}_{20}\text{N}_5\text{O}$ [$\text{M} + \text{H}^+$] 370.1662, found 370.1663; IR (KBr) ν_{max} 3083, 3030, 2931, 1637, 1374, 1247, 808 cm^{-1} .

3-Benzyl-7-(2-methoxybenzylidene)-7,8-dihydro-3H-imidazo[2,1-*ij*]purine (3m). Green oil (13.2 mg, 11.9%): ^1H NMR (400 MHz, CDCl_3) δ 8.62 (s, 1H), 7.91 (s, 1H), 7.31 (s, 4H), 7.26 (s, 3H), 7.18 (s, 1H), 6.90 (t, $J = 8.2$ Hz, 2H), 5.43 (s, 2H), 4.20 (s, 2H), 3.82 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 157.3, 147.7, 141.0, 140.3, 139.1, 135.5, 134.8, 130.8, 129.0, 128.4, 127.8, 127.6, 127.5, 123.3, 120.6, 110.5, 107.5, 55.4, 47.7, 29.6; HRMS calcd for $\text{C}_{22}\text{H}_{20}\text{N}_5\text{O}$ [$\text{M} + \text{H}^+$] 370.1662, found 370.1660; IR (KBr) ν_{max} 3079, 3017, 2832, 1641, 1371, 1244, 723 cm^{-1} .

3-Benzyl-7-(4-methoxybenzyl)-3H-imidazo[2,1-*ij*]purine (2n). Yellow oil (82.8 mg, 74.8%): ^1H NMR (400 MHz, CDCl_3) δ 8.45 (s, 1H), 7.94 (s, 1H), 7.44 (s, 1H), 7.31–7.27 (m, 3H), 7.20–7.23 (m, 2H), 7.11 (d, $J = 7.6$ Hz, 2H), 6.82 (d, $J = 8.8$ Hz, 2H), 5.41 (s, 2H), 4.24 (s, 2H), 3.76 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 158.6, 135.5, 133.8, 129.2, 128.9, 128.3, 127.5, 114.3, 55.2, 47.7, 29.6; HRMS calcd for $\text{C}_{22}\text{H}_{20}\text{N}_5\text{O}$ [$\text{M} + \text{H}^+$] 370.1662, found 370.1662; IR (KBr) ν_{max} 3079, 2977, 2927, 1639, 1246, 703, 525 cm^{-1} .

3-Benzyl-7-(4-methoxybenzylidene)-7,8-dihydro-3H-imidazo[2,1-*ij*]purine (3n). Yellow solid (11.3 mg, 10.2%): mp 175–177 °C; ^1H NMR (400 MHz, CDCl_3) δ 8.65 (s, 1H), 7.93 (s, 1H), 7.34–7.25 (m, 7H), 7.19 (s, 1H), 6.86 (d, $J = 8.4$ Hz, 2H), 5.45 (s, 2H), 4.13 (s, 2H), 3.78 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 158.1, 135.5, 134.9, 131.1, 130.1, 129.0, 128.3, 127.5, 113.8, 55.2, 47.7, 34.7; HRMS calcd for $\text{C}_{22}\text{H}_{20}\text{N}_5\text{O}$ [$\text{M} + \text{H}^+$] 370.1662, found 370.1663; IR (KBr) ν_{max} 3082, 2984, 2918, 1645, 1238, 725 cm^{-1} .

3-Benzyl-7-(4-ethoxybenzyl)-3H-imidazo[2,1-*ij*]purine (2o). Brown oil (93.1 mg, 81%): ^1H NMR (400 MHz, CDCl_3) δ 8.46 (s, 1H), 7.93 (s, 1H), 7.45 (s, 1H), 7.32 (s, 3H), 7.25 (s, 2H), 7.11 (d, $J = 4.4$ Hz, 2H), 6.82 (s, $J = 8.0$ Hz, 2H), 5.43 (s, 2H), 4.26 (s, 2H), 3.99 (q, $J = 7.2$ Hz, 2H), 1.39 (t, $J = 7.2$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 158.1, 141.8, 140.3, 138.9, 135.5, 133.8, 132.3, 129.2, 129.0, 128.4, 127.9, 127.6, 123.7, 122.4, 114.9, 63.5, 47.8, 29.7, 14.8; HRMS calcd for $\text{C}_{23}\text{H}_{21}\text{N}_5\text{NaO}$ [$\text{M} + \text{Na}^+$] 406.1638, found 406.1647; IR (KBr) ν_{max} 3098, 3010, 2983, 1641, 1375, 1241, 705, 519 cm^{-1} .

3-Benzyl-7-(4-ethoxybenzylidene)-7,8-dihydro-3H-imidazo[2,1-*ij*]purine (3o). Yellow solid (8.1 mg, 7%): mp 146–148 °C; ^1H NMR (400 MHz, CDCl_3) δ 8.63 (s, 1H), 7.91 (s, 1H), 7.33–7.30 (m, 3H), 7.28 (s, 2H), 7.26–7.24 (m, 2H), 7.12 (s, 1H), 6.84 (d, $J = 8.8$ Hz, 2H), 5.44 (s, 2H), 4.12 (s, 2H), 4.01 (q, $J = 7.20$ Hz, 2H), 1.40 (t, $J = 7.2$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 157.6, 148.6, 141.3, 140.3, 139.2, 135.5, 134.9, 131.0, 130.1, 129.0, 128.4, 127.5, 123.3, 114.5, 107.5, 63.4, 47.7, 34.7, 14.9; HRMS calcd for $\text{C}_{23}\text{H}_{21}\text{N}_5\text{NaO}$ [$\text{M} + \text{Na}^+$] 406.1638, found 406.1641; IR (KBr) ν_{max} 3095, 3013, 2981, 1647, 1371, 1238, 723 cm^{-1} .

3-Benzyl-7-(3-chlorobenzyl)-3H-imidazo[2,1-*ij*]purine (2p). Brown oil (42.6 mg, 38.1%): ^1H NMR (400 MHz, CDCl_3) δ 8.45 (s, 1H), 7.97 (s, 1H), 7.50 (s, 1H), 7.35–7.31 (m, 3H), 7.24–7.23 (m, 5H), 7.22 (s, 1H), 7.11–7.09 (m, 1H), 5.45 (s, 2H), 4.32 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 140.5, 139.0, 138.2, 135.3, 134.9, 133.4, 132.7, 130.2, 129.0, 128.4, 128.3, 127.5, 126.3, 123.6, 121.0, 47.8, 30.1; HRMS calcd for $\text{C}_{21}\text{H}_{16}\text{ClN}_5\text{Na}$ [$\text{M} + \text{Na}^+$] 396.0986, found 396.0981; IR (KBr) ν_{max} 3128, 3063, 2922, 1642, 1371, 715 cm^{-1} .

3-Benzyl-7-(3-chlorobenzylidene)-7,8-dihydro-3H-imidazo[2,1-*ij*]purine (3p). Light yellow solid (48.1 mg, 42.9%): mp 210–213 °C; ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{SO}$) δ 9.27 (s, 1H), 8.53 (s, 1H), δ 7.90 (s, 1H), 7.42 (s, 1H), 7.35–7.28 (m, 8H), 5.55 (s, 2H), 4.14 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 146.9, 141.1, 140.4, 139.2, 135.4, 134.9, 134.1, 129.7, 129.1, 129.0, 128.4, 127.5, 127.3, 126.6, 123.3, 107.7, 47.7, 35.1; HRMS calcd for $\text{C}_{21}\text{H}_{17}\text{ClN}_5$ [$\text{M} + \text{H}^+$] 374.1167, found 374.1164; IR (KBr) ν_{max} 3063, 2925, 2854, 1641, 1376, 698 cm^{-1} .

3-Benzyl-7-(4-chlorobenzyl)-3H-imidazo[2,1-*ij*]purine (2q). Brown oil (44.3 mg, 39.6%): ^1H NMR (400 MHz, CDCl_3) δ 8.43 (s, 1H), 7.97 (s, 1H), 7.48 (s, 1H), 7.34–7.27 (m, 7H), 7.16 (s, 1H), 7.14 (s,

1H), 5.44 (s, 2H), 4.30 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 135.4, 134.7, 133.5, 133.1, 129.6, 129.2, 129.0, 128.4, 127.6, 47.8, 30.9; HRMS calcd for $\text{C}_{21}\text{H}_{17}\text{ClN}_5$ [$\text{M} + \text{H}^+$] 374.1167, found 374.1164; IR (KBr) ν_{max} 3136, 3066, 3030, 1638, 1371, 725 cm^{-1} .

3-Benzyl-7-(4-chlorobenzylidene)-7,8-dihydro-3H-imidazo[2,1-*ij*]purine (3q). Light yellow solid (54.2 mg, 48.4%): mp 205–207 °C; ^1H NMR (400 MHz, CDCl_3) δ 8.59 (s, 1H), 7.85 (s, 1H), 7.26–7.17 (m, 1H), 8.19 (s, 1H), 5.37 (s, 2H), 4.06 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 147.3, 141.4, 140.4, 139.3, 137.6, 135.5, 134.9, 132.2, 130.4, 129.0, 128.6, 128.4, 127.6, 123.3, 107.6, 47.8, 34.9; HRMS calcd for $\text{C}_{21}\text{H}_{17}\text{ClN}_5$ [$\text{M} + \text{H}^+$] 374.1167, found 374.1167; IR (KBr) ν_{max} 3136 3077, 2934, 1634, 1371, 692 cm^{-1} .

3-Benzyl-7-(4-fluorobenzyl)-3H-imidazo[2,1-*ij*]purine (2r). Brown oil (34.6 mg, 32.3%): ^1H NMR (400 MHz, CDCl_3) δ 8.36 (s, 1H), 7.89 (s, 1H), 7.25–7.20 (m, 7H), 7.34–7.27 (m, 7H), 7.09 (s, 1H), 7.07 (s, 1H), 5.36 (s, 2H), 4.23 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 135.4, 134.7, 133.1, 129.6, 129.2, 129.0, 128.4, 127.6, 47.8, 29.9; HRMS calcd for $\text{C}_{21}\text{H}_{17}\text{FN}_5$ [$\text{M} + \text{H}^+$] 358.1463, found 358.1456; IR (KBr) ν_{max} 3084, 2937, 2839, 1638, 1375, 1240, 727 cm^{-1} .

3-Benzyl-7-(4-fluorobenzylidene)-7,8-dihydro-3H-imidazo[2,1-*ij*]purine (3r). Brown solid (45.8 mg, 42.8%): mp 144–146 °C; ^1H NMR (400 MHz, CDCl_3) δ 8.67 (s, 1H), 7.92 (s, 1H), 7.31–7.24 (m, 8H), 6.98 (t, $J = 8.4$ Hz, 2H), 5.44 (s, 2H), 4.14 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 162.8, 160.4, 147.7, 141.4, 140.4, 139.2, 135.5, 134.9, 134.8, 134.8, 130.5, 130.5, 129.0, 128.4, 127.5, 123.3, 115.3, 115.1, 107.6, 47.7, 34.7; HRMS calcd for $\text{C}_{21}\text{H}_{16}\text{FN}_5\text{Na}$ [$\text{M} + \text{Na}^+$] 380.1282, found 380.1273; IR (KBr) ν_{max} 3074, 3014, 2917, 2865, 1638, 1375, 695 cm^{-1} .

3-Benzyl-7-(thiophen-2-ylmethyl)-3H-imidazo[2,1-*ij*]purine (2s). Brown oil (35.9 mg, 34.7%): ^1H NMR (400 MHz, CDCl_3) δ 8.68 (s, 1H), 7.93 (s, 1H), 7.37–7.33 (m, 5H), 7.26 (s, 1H), 7.18 (d, $J = 4.8$ Hz, 2H), 7.00 (s, 1H), 6.96 (s, 1H), 5.45 (s, 2H), 4.40 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 147.2, 141.0, 140.4, 139.2, 135.4, 134.9, 129.0, 128.4, 127.5, 126.8, 125.9, 124.0, 107.6, 47.8, 29.7; HRMS calcd for $\text{C}_{19}\text{H}_{15}\text{N}_5\text{NaS}$ [$\text{M} + \text{Na}^+$] 368.0940, found 368.0947; IR (KBr) ν_{max} 3071, 2923, 1640, 1373, 1350, 688, 647 cm^{-1} .

3-Benzyl-7-(thiophen-2-ylmethylene)-7,8-dihydro-3H-imidazo[2,1-*ij*]purine (3s). Brown solid (43.8 mg, 42.4%): mp 165–167 °C; ^1H NMR (400 MHz, CDCl_3) δ 8.69 (s, 1H), 7.94 (s, 1H), 7.38 (s, 1H), 7.34 (s, 3H), 7.27 (s, 2H), 7.19 (d, $J = 4.0$ Hz, 1H), 7.00 (s, 1H), 6.97 (s, 1H), 5.46 (s, 2H), 4.41 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 147.2, 141.1, 140.4, 135.4, 135.0, 129.0, 128.4, 127.6, 126.9, 125.9, 124.1, 123.3, 107.6, 47.8, 29.8; HRMS calcd for $\text{C}_{19}\text{H}_{15}\text{N}_5\text{NaS}$ [$\text{M} + \text{Na}^+$] 368.0940, found 368.0944; IR (KBr) ν_{max} 3071, 2924, 1640, 1373, 1345, 725, 647 cm^{-1} .

■ ASSOCIATED CONTENT

Supporting Information

Copies of ^1H and ^{13}C NMR spectra of compounds **2a–2s** and **3i–3s**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

■ AUTHOR INFORMATION

Corresponding Authors

*Fax: (+86) 373-3329276. E-mail: quguir@sina.com.

*E-mail: guohm518@hotmail.com.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

We are grateful for financial support from the National Natural Science Foundation of China (Grant Nos. 21172059, 21272059, 21202039, and 21372066), Excellent Youth Foundation of Henan Scientific Committee (No. 114100510012), the Program for Innovative Research Team from the University of Henan Province (2012IRTSTHN006), the Program for Changjiang Scholars and Innovative Research

Team in the University (IRT1061), the Program for Science & Technology Innovation Talents in Universities of Henan Province (No. 13HASTIT013), and the Foundation for University Young Key Teacher by Henan Province of China (No. 2011GGJS-132).

REFERENCES

- (1) (a) Wolfe, A. E.; O'Brien, P. J. *Biochemistry* **2009**, *48*, 11357. (b) Liu, M.; Xu, M.; Lee, J. K. *J. Org. Chem.* **2008**, *73*, 5907. (c) Mishina, Y.; Yang, C.-G.; He, C. *J. Am. Chem. Soc.* **2005**, *127*, 14594. (d) Hang, B.; Singer, B.; Margison, G. P.; Elder, R. H. *Proc. Natl. Acad. Sci. U. S. A.* **1997**, *94*, 12869. (e) Chenna, A.; Hang, B.; Rydberg, B.; Kim, E.; Pongracz, K.; Bodell, W. J.; Singer, B. *Proc. Natl. Acad. Sci. U. S. A.* **1995**, *92*, 5890. (f) Rydberg, B.; Dosanjh, M. K.; Singer, B. *Proc. Natl. Acad. Sci. U. S. A.* **1991**, *88*, 6839.
- (2) (a) Chen, H.-J. C.; Chiang, L.-C.; Tseng, M.-C.; Zhang, L. L.; Ni, J.; Chung, F.-L. *Chem. Res. Toxicol.* **1999**, *12*, 1119. (b) Ghissassi, F. E.; Barbin, A.; Nair, J.; Bartsch, H. *Chem. Res. Toxicol.* **1995**, *8*, 278. (c) Kuśmierk, J. T.; Jensen, D. E.; Spengler, S. J.; Stolarski, R.; Singer, B. *J. Org. Chem.* **1987**, *52*, 2374. (d) Jefferson, J. R.; Hunt, J. B.; Jamieson, G. A. *J. Med. Chem.* **1987**, *30*, 2013. (e) Gruber, B. A.; Leonard, N. J. *Proc. Natl. Acad. Sci. U. S. A.* **1975**, *72*, 3966. (f) Kobori, A.; Morita, J.; Ikeda, M.; Yamayoshi, A.; Murakami, A. *Bioorg. Med. Chem. Lett.* **2009**, *19*, 3657. (g) Secrist, J. A., III; Barrio, J. R.; Leonard, N. J.; Villar-Palasi, C.; Gilman, A. G. *Science* **1972**, *177*, 279.
- (3) Yip, K.-F.; Tsou, K.-C. *J. Org. Chem.* **1975**, *40*, 1066.
- (4) Kochetkov, N. K.; Shibaev, V. N.; Kost, A. A. *Tetrahedron Lett.* **1971**, *12*, 1993.
- (5) For selected examples for the synthesis of nitrogen-containing heterocycles via transition metal-catalyzed coupling reactions, see: (a) Zhu, Y.-P.; Jia, F.-C.; Liu, M.-C.; Wu, A.-X. *Org. Lett.* **2012**, *14*, 4414. (b) Sang, P.; Xie, Y.; Zou, J.; Zhang, Y. *Org. Lett.* **2012**, *14*, 3894. (c) Oda, Y.; Hirano, K.; Satoh, T.; Miura, M. *Org. Lett.* **2012**, *14*, 664. (d) Nanjo, T.; Tsukano, C.; Takemoto, Y. *Org. Lett.* **2012**, *14*, 4270. (e) Hikawa, H.; Ino, Y.; Suzuki, H.; Yokoyama, Y. *J. Org. Chem.* **2012**, *77*, 7046. (f) Liao, Q.; Zhang, L.; Li, S.; Xi, C. *Org. Lett.* **2011**, *13*, 228. (g) Priebbenow, D. L.; Henderson, L. C.; Pfeffer, F. M.; Stewart, S. G. *J. Org. Chem.* **2010**, *75*, 1787. (h) Bryan, C. S.; Lautens, M. *Org. Lett.* **2008**, *10*, 4633.
- (6) For selected examples that propargyl substrates were used to construct heterocyclic compounds, see: (a) Yan, W.; Ye, X.; Weise, K.; Petersen, J. L.; Shi, X. *Chem. Commun.* **2012**, *48*, 3521. (b) Xu, T.; Liu, G. *Org. Lett.* **2012**, *14*, 5416. (c) Wang, Y.; Ready, J. M. *Org. Lett.* **2012**, *14*, 2308. (d) Polindara-García, L. A.; Miranda, L. D. *Org. Lett.* **2012**, *14*, 5408. (e) Li, S.; Li, Z.; Yuan, Y.; Peng, D.; Li, Y.; Zhang, L.; Wu, Y. *Org. Lett.* **2012**, *14*, 1130. (f) Gronnier, C.; Odabachian, Y.; Gagosz, F. *Chem. Commun.* **2011**, *47*, 218. (g) Efe, C.; Lykakis, I. N.; Stratakis, M. *Chem. Commun.* **2011**, *47*, 803. (h) Kim, I.; Kim, K. *Org. Lett.* **2010**, *12*, 2500. (i) Donets, P. A.; Hecke, K. V.; Meervelt, L. V.; Van der Eycken, E. V. *Org. Lett.* **2009**, *11*, 3618. (j) Aponick, A.; Li, C.-Y.; Malinge, J.; Marques, E. F. *Org. Lett.* **2009**, *11*, 4624. (k) Cacchi, S.; Fabrizi, G.; Filisti, E. *Org. Lett.* **2008**, *10*, 2629. (l) Sandelier, M. J.; DeShong, P. *Org. Lett.* **2007**, *9*, 3209. (m) Vachhani, D. D.; Modha, S. G.; Sharma, A.; Van der Eycken, E. V. *Tetrahedron* **2013**, *69*, 359.
- (7) (a) Guo, H.-M.; Xia, C.; Niu, H.-Y.; Zhang, X.-T.; Kong, S.-N.; Wang, D.-C.; Qu, G.-R. *Adv. Synth. Catal.* **2011**, *353*, 53. (b) Niu, H.-Y.; Yuan, T.-F.; Qu, G.-R.; Li, D.-Y.; Mao, R.-Z.; Jin, X.; Yang, X.-N.; Guo, H.-M. *Chin. J. Org. Chem.* **2011**, *10*, 1613. (c) Guo, H.-M.; Yuan, T.-F.; Niu, H.-Y.; Liu, J.-Y.; Mao, R.-Z.; Li, D.-Y.; Qu, G.-R. *Chem.—Eur. J.* **2011**, *17*, 4095. (d) Meng, G.; Niu, H.-Y.; Qu, G.-R.; Fossey, J. S.; Li, J.-P.; Guo, H.-M. *Chem. Commun.* **2012**, *48*, 9601. (e) Qu, G.-R.; Liang, L.; Niu, H.-Y.; Rao, W.-H.; Guo, H.-M.; Fossey, J. S. *Org. Lett.* **2012**, *14*, 4494. (f) Li, J.-P.; Huang, Y.; Xie, M.-S.; Qu, G.-R.; Niu, H.-Y.; Wang, H.-X.; Qin, B.-W.; Guo, H.-M. *J. Org. Chem.* **2013**, *78*, 12629. (g) Wang, D.-C.; Niu, H.-Y.; Xie, M.-S.; Qu, G.-R.; Wang, H.-X.; Guo, H.-M. *Org. Lett.* **2014**, *16*, 262. (h) Xie, M.-S.; Chu, Z.-L.; Niu, H.-Y.; Qu, G.-R.; Guo, H.-M. *J. Org. Chem.* **2014**, *79*, 1093.
- (8) Guo, H.-M.; Wu, S.; Niu, H.-Y.; Song, G.; Qu, G.-R. *Chemical Synthesis of Acyclic Nucleosides in Chemical Synthesis of Nucleoside Analogues 3*; Pedro, M., Ed.; John Wiley & Sons: New York, 2013; pp 103–162.
- (9) (a) Ma, D.; Cai, Q. *Acc. Chem. Res.* **2008**, *41*, 1450. (b) Ueda, S.; Nagasawa, H. *J. Am. Chem. Soc.* **2009**, *131*, 15080. (c) Fructos, M. R.; Álvarez, E.; Díaz-Requejo, M. M.; Pérez, P. J. *J. Am. Chem. Soc.* **2010**, *132*, 4600. (d) Nakamura, I.; Zhang, D.; Terada, M. *J. Am. Chem. Soc.* **2010**, *132*, 7884. (e) Wan, C.; Zhang, J.; Wang, S.; Fan, J.; Wang, Z. *Org. Lett.* **2010**, *12*, 2338. (f) Qiu, G.; Hu, Y.; Ding, Q.; Peng, Y.; Hu, X.; Wu, J. *Chem. Commun.* **2011**, *47*, 9708. (g) Liu, G.; Liu, H.; Qiu, G.; Pu, S.; Wu, J. *Chem. Commun.* **2012**, *48*, 7049. (h) Yan, J.; Zhou, F.; Qin, D.; Cai, T.; Ding, K.; Cai, Q. *Org. Lett.* **2012**, *14*, 1262.
- (10) (a) Liu, K.; Wu, X.; Jennifer Kan, S. B.; Shirakawa, S.; Maruoka, K. *Chem.—Asian. J.* **2013**, *8*, 3214. (b) Elsner, P.; Bernardi, L.; Salla, G. D.; Overgaard, J.; Jørgensen, K. A. *J. Am. Chem. Soc.* **2008**, *130*, 4897.
- (11) Borrmann, T.; Abdelrahman, A.; Volpini, R.; Lambertucci, C.; Alksnis, E.; Gorzalka, S.; Knospe, M.; Schiedel, A. C.; Cristalli, G.; Muller, C. E. *J. Med. Chem.* **2009**, *52*, 5974.
- (12) Chinchilla, R.; Nájera, C. *Chem. Rev.* **2007**, *107*, 874.